(19) World Intellectual Property Organization International Bureau



(43) International Publication Date 22 February 2001 (22.02.2001)

PCT

(10) International Publication Number WO 01/12540 A1

(51) International Patent Classification7:

. . .

(21) International Application Number: PCT/US00/22421

(22) International Filing Date: 15 August 2000 (15.08.2000)

(25) Filing Language:

English

C01B 3/38

(26) Publication Language:

English

(30) Priority Data:

09/375,615

17 August 1999 (17.08.1999) U

- (71) Applicant: BATTELLE MEMORIAL INSTITUTE [US/US]; Pacific Northwest Division, Intellectual Property Services, P.O. Box 999, Richland, WA 99352 (US).
- (72) Inventors: WANG, Yong: 2884 Troon Ct., Richland, WA 99352 (US). VANDERWIEL, David, P.: 2550 Duportail St., G-241. Richland, WA 99352 (US). TONKOVICH, Anna, Lee, Y.: 1604 North 37th Place North, Pasco, WA 99301 (US).

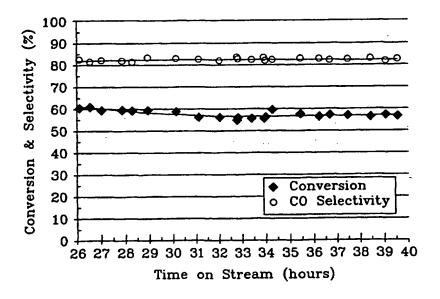
- (74) Agent: MAY, Stephen, R.: Battelle Memorial Institute. Pacific Northwest Division, Intellectual Property Services. P.O. Box 999, MSIN: K1-53, Richland, WA 99352 (US).
- (81) Designated States (national): AE. AG. AL. AM. AT. AU. AZ. BA. BB. BG. BR. BY. BZ. CA. CH. CN. CR. CU. CZ. DE. DK. DM. DZ. EE. ES. FI. GB. GD. GE. GH. GM. HR. HU. ID. IL. IN, IS, JP. KE. KG. KP. KR. KZ. LC. LK. LR. LS. LT. LU. LV. MA. MD. MG. MK. MN. MW. MX. MZ. NO. NZ. PL. PT. RO. RU. SD. SE. SG. SI. SK, SL. TJ. TM. TR. TT. TZ. UA. UG. UZ. VN. YU. ZA. ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).

Published:

With international search report.

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: A METHOD AND CATALYST STRUCTURE FOR STEAM REFORMING OF A HYDROCARBON



(57) Abstract: The present invention includes an improvement to the existing method of steam reforming of hydroc abon, wherein the improvement comprises: the flowing is at a rate providing a residence time less than about 0.1 sec resulting in obtaining product formation yield or amount that is the same or greater compared to product formation at a longer residence time. Another improvement of the present invention is operation at a steam to carbon ratio that is substantially stoichiometric and maintaining activity of the supported catalyst. The present invention also includes a catalyst structure for steam reforming of a hydrocarbon.

540 A1

A METHOD AND CATALYST STRUCTURE FOR STEAM REFORMING OF A HYDROCARBON

FIELD OF THE INVENTION

The present invention is a method and catalyst structure for steam reforming of a hydrocarbon.

10

20

25

30

BACKGROUND OF THE INVENTION

Steam reforming of hydrocarbons is commonly used for feedstock production for carbon-monoxide hydrogenation (Fischer-Tropsch synthesis), methanol synthesis and hydrogen production. Steam reforming is done commercially by flowing a mixture of steam and the hydrocarbon past a supported catalyst having an alumina support and a catalyst metal thereon, and reacting the mixture at a temperature from about 600 °C to about 1000 °C, forming at least one product. Research has been done with the catalyst metal on a spinel support. Residence times are typically on the order of seconds and steam tocarbon ratio greater than about 2.5. For steam to carbon ratio less than 2.5, catalyst activity is generally degraded after hours to days due to coke formation and the supported catalyst must be refreshed or replaced.

The rate of supported catalyst activity degradation has been reduced by use of excess steam (steam to carbon ratio greater than 2.5). Excess steam, however, requires excess thermal energy and results in large system pressure drop. Using less steam results in faster degradation of catalyst activity because of coking from the hydrocarbon(s).

Hence, there is a need for a method of steam reforming of a hydrocarbon that provides greater product yield and permits using less steam and maintaining catalytic activity of the catalyst.

WO 01/12540 PCT/US00/22421

SUMMARY OF THE INVENTION

The present invention includes an improvement to the existing method of steam reforming of hydrocarbon, wherein the improvement comprises:

the flowing is at a rate providing a residence time less than about 0.1 sec resulting in obtaining product formation yield or amount that is the same or greater compared to product formation at a longer residence time.

Another improvement of the present invention is operation at a steam to carbon ratio that is substantially stoichiometric and maintaining activity of the supported catalyst.

The present invention also includes a catalyst structure for steam reforming of a hydrocarbon. The catalyst structure has

- (a) a first porous structure with a first pore surface area and a first pore size of at least about 0.1 μm;
- (b) a porous interfacial layer that is a spinel with a second pore surface area and a second pore size less than the first pore size, the porous interfacial layer having a thickness less than 4 mm placed upon the first pore surface area;
- (c) a steam reforming catalyst selected from the group consisting of rhodium, iridium, nickel, palladium, platinum, carbide of group VIb and combinations thereof placed upon the second pore surface area.

It is an object of the present invention to provide a method of steam reforming of hydrogen with a residence time of less than about 0.1 sec.

It is an object of the present invention to provide a catalyst structure with a porous interfacial layer of spinel.

The subject matter of the present invention is particularly pointed out and distinctly claimed in the concluding portion of this specification. However, both the organization and method of operation, together with further advantages and objects thereof, may best be understood by reference to the following description taken in connection with accompanying drawings wherein like reference characters refer to like elements.

5

10

15

20

25

30

BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 is a graph of conversion and selectivity versus temperature.

FIG. 2 is a graph of conversion and selectivity versus time.

DESCRIPTION OF THE PREFERRED EMBODIMENT(S)

The present invention includes a method for steam reforming of a hydrocarbon having the steps of flowing a mixture of steam and the hydrocarbon past a supported catalyst having a support and a catalyst metal thereon. The mixture is reacted at a temperature from about 600 °C to about 1000 °C forming at least one product. The improvement of the present invention is using a spinel support and flowing the mixture at a rate providing a residence time less than about 0.1 sec and obtaining product formation that is the same or greater than that obtained at longer residence times.

Also, under the previously described conditions, catalytic activity is degraded when the steam to carbon ratio is substantially stoichiometric. Another improvement of the present invention realized by flowing the mixture at a rate providing a residence time less than about 0.1 sec is maintaining activity of the spinel supported catalyst beyond 6 hours without degradation by coking even for substantially stoichiometric steam to carbon ratio. Substantially stoiciometric is a steam to carbon content ratio greater than about 0.9 and less than about 2.5, preferably from about 0.98 to about 2.

The supported catalyst may be in the form of a powder of non-porous particles, porous solid and combinations thereof.

Hydrocarbon includes oxygenates, alkanes, alkenes, alkynes, branched isomers, aromatics, saturated and unsaturated hydrocarbons and combinations thereof including fuels such as gasoline, kerosine, diesel, JP-8.

30 Example 1

An experiment was conducted to demonstrate the present invention. The supported catalyst was spinel of a gamma-alumina (γ -Al₂O₃) support with a magnesia (MgO) passivation layer and rhodium oxide (Rh₂O₃). The approximate

20

25

composition was about 15 wt% Rh₂O₃, about 5 wt% MgO, and about 80 wt% γ -Al₂O₃. The supported catalyst was prepared by (1) calcining a high surface area γ -Al₂O₃ at 500 °C for 5 hours; (2) impregnating the γ -Al₂O₃ with MgO using the incipient wetness technique with a solution of magnesium nitrate; and obtaining an MgO modified γ -Al₂O₃ support; (3) drying the modified support at 110 °C for 4 hours followed by (4) a second calcination at 900 °C for 2 hours; (5) impregnating the modified support with Rh₂O₃ with the incipent wetness technique from a rhodium nitrate solution; (6) followed by a final drying 110 °C for 4 hours and a (7) a final calcination at 500 °C for 3 hours to obtain a powder of the supported catalyst.

A microreactor was constructed of a quartz tube with 4mm ID and 6.35 mm OD. About 0.2 g of powder of supported catalyst was placed in the microreactor in a packed bed arrangement.

Reactants were steam and methane in a steam to carbon ratio of approximately 1 which is stoichiometric within measurement uncertainty.

Reactants were flowed through the reactor at temperatures from 650 °C to 900 °C.

Results are shown in **FIG. 1** for a steam to carbon ratio of 3 with conversion ranging from about 52% to 95% with increasing temperature and selectivity ranging from 22% to 70%.

Results in FIG. 2 are for a steam to carbon ratio of 1 at 900 °C over 40 hours. No degradation of the supported catalyst was observed. Electron microscopic examination after testing revealed no coke deposition and BET measurements detected no significant loss in surface area.

25

10

15

20

CLOSURE

While a preferred embodiment of the present invention has been shown and described, it will be apparent to those skilled in the art that many changes and modifications may be made without departing from the invention in its broader aspects. The appended claims are therefore intended to cover all such changes and modifications as fall within the true spirit and scope of the invention.

CLAIMS

We claim:

5

1. A method for steam reforming of a hydrocarbon having the steps of flowing a mixture of steam and said hydrocarbon past a supported catalyst having a support and a catalyst metal thereon, and reacting said mixture at a temperature from about 600 °C to about 1000 °C forming at least one product; wherein the improvement comprises:

said support is a spinel support; and

said flowing is at a rate providing a residence time less than about 0.1 sec and obtaining said forming the same or greater compared to said forming at a longer residence time.

15

20

25

- 2. The method as recited in claim 1, wherein said mixture has a steam to carbon ratio less than 2.5, said improvement maintaining activity of said supported catalyst beyond 6 hours.
- 3. The method as recited in claim 1, wherein said spinel support controls acidity of said supported catalyst.
 - 4. A method for steam reforming of a hydrocarbon having the steps of flowing a mixture of steam and said hydrocarbon having a steam to carbon ratio that is substantially stoichiometric past a supported catalyst having a support and a catalyst metal thereon, and reacting said mixture at a temperature from about 600 °C to about 1000 °C forming at least one product and degrading catalytic activity of said supported catalyst; wherein the improvement comprises:

said support is a spinel support; and

30

said flowing is at a rate providing a residence time less than about 0.1 sec and maintaining activity of said supported catalyst for said steam to carbon ratio less than 2.5.

5

10

15

20

- 5. The method as recited in claim 4, wherein said support is spinel that controls acidity of said supported catalyst.
- 6. The method as recited in claim 4, wherein said steam to carbon ratio is greater than about 0.9 and less than about 2.5.
- 7. The method as recited in claim 4, wherein said supported catalyst is on a porous substrate.

8. A catalyst structure for steam reforming of a hydrocarbon, comprising:

- (a) a first porous structure with a first pore surface area and a first pore size of at least about 0.1 μm ;
 - (b) a buffer layer upon said first pore surface area;
- (c) a porous interfacial layer that is a spinel with a second pore surface area and a second pore size less than said first pore size, said porous interfacial layer having a thickness less than 4 mm placed upon said buffer layer;
- (d) a steam reforming catalyst selected from the group consisting of rhodium, iridium, nickel, palladium, platinum, carbide of group IVb and combinations thereof placed upon said second pore surface area.
- 9. The catalyst structure as recited in claim 8, wherein said carbide is selected from the group of tungsten carbide, molybdenum carbide and combinations thereof.

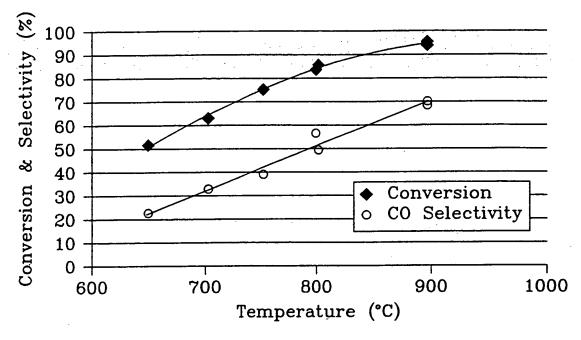


Fig. 1

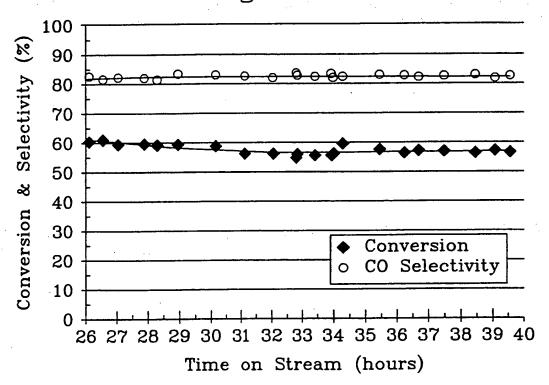


Fig. 2

INTERNATIONAL SEARCH REPORT

Interna 31 Application No PCT/US 00/22421

A. CLASSIF	CO183/38			
	International Patent Classification (IPC) or to both national classifica	tion and IPC		
B. FIELDS !	SEARCHED currentation searched (classification system followed by classification	on symbols)		
IPC 7				
•				
	ion searched other than minimum documentation to the extent that su			
Electronic da	ata base consulted during the international search (name of data bas	se and, where practical, search terms used		
EPO-In	ternal, WPI Data			
C. DOCUME	ENTS CONSIDERED TO BE RELEVANT			
Category *	Citation of document, with indication, where appropriate, of the rele	evant passages	Relevant to claim No.	
	TOYO ENGINEEDING	0000)	1	
Α	FR 2 105 253 A (TOYO ENGINEERING 28 April 1972 (1972-04-28)	CORP	1	
	page 8, line 18 -page 9, line 36			
	US 4 088 608 A (ITO NOBUEI ET AL)	\	1	
A	9 May 1978 (1978-05-09)		_	
	claim 1; example 1			
A	GB 1 003 147 A (CHEMICAL CONSTRUC	CTION	1	
^	CORPORATION) 2 September 1965 (19	965-09-02)		
	page 2, line 95 -page 3, line 39	v		
l _A	US 5 679 614 A (BANGALA DENIS NGC	DY ET AL)	1	
	21 October 1997 (1997-10-21)	. 16.		
	column 9, line 60 -column 10, lir example 1	ie 10,		
ļ				
Funt	ther documents are listed in the continuation of box C.	X Patent family members are listed	I in annex.	
Special ca	ategones of cited documents :	"T" later document published after the int or priority date and not in conflict with	emational filing date	
"A" docum	ent defining the general state of the lart which is not dered to be of particular relevance	cited to understand the principle or tr	neory underlying the	
	document but published on or after the international	"X" document of particular relevance; the cannot be considered novel or cannot	t be considered to	
"L" docum	ent which may throw doubts on prionty claim(s) or his cated to establish the publication date of another	involve an inventive step when the di "Y" document of particular relevance; the	ocument is taken alone claimed invention	
citatio	on or other special reason (as specified) ment referring to an oral disclosure, use, exhibition or	cannot be considered to involve an in	nventive step when the lore other such docu-	
other	means sent published prior to the international filling date but	ments, such combination being obvious the art.		
later t	than the priority date claimed	*&* document member of the same paten Date of mailing of the international se		
Date of the	e actual completion of the international search	<u>-</u>		
1	14 November 2000	21/11/2000		
Name and	mailing address of the ISA	Authorized officer		
	European Patent Office, P.B. 5816 Patentaan 2 NL – 2280 HV Rijswyk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl.	Clement, J-P		
	Fax: (+31-70) 340-3016	Clement, 5-r		

Form PCT/ISA/210 (second sheet) (July 1992)

1

INTERNATIONAL SEARCH REPORT

Information on patent family members

"Interna 1 Application No PCT/US 00/22421

Patent document cited in search report			Publication date	Patent family member(s)		Publication date
FR	2105253	Α	28-04-1972	AU	462426 B	26-06-1975
		•		AU	3310471 A	08-03-1973
				CA	949325 A	18-06-1974
				DE	2143608 A	09-03-1972
	$(\tau,\tau) = (1/\tau)^{-1/\tau}$			NL	7112223 A,B,	07-03-1972
				SU	526281 A	25-08-1976
	,	, .	•	US	3957681 A	18-05-1976
				US	4025457 A	24-05-1977
US	4088608	Α	09-05-1978	 ЈР	52074591 A	22-06-1977
•	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			DE	2657126 A	30-06-1977
GB	1003147	Α		NONE		
	5679614	A	21-10-1997	 CA	2114965 A	05-08-1995

Form PCT/ISA/210 (patent family annex) (July 1992)